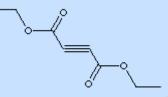
DIETHYL ACETYLENEDICARBOXYLATE

SYNONYMS

1,2-Bis(ethoxycarbonyl)acetylene; 1,2-Bis(ethoxycarbonyl)ethyne; Acetylenedicarboxylic acid, diethyl ester; Bis(ethoxycarbonyl)acetylene; Di(carboethoxy)acetylene; Diethyl 2-butynedioate; Diethyl acetylenedicarboxylic acid; Diethyl butynedioate; Diethyl ethynedicarboxylate; Ethyl acetylenedicarboxylate;

PRODUCT IDENTIFICATION

CAS RN EINECS RN FORMULA MOL WEIGHT 762-21-0 212-098-4 C₂H₅OCOC≡CCOOC₂H₅ 170.16



PHYSICAL AND CHEMICAL PROPERTIES

colorless to light yellow liquid PHYSICAL STATE **MELTING POINT** 1 - 2 C **BOILING POINT** 107 - 110 C at 11 mmHg DENSITY 1.06 - 1.07 SOLUBILITY IN WATER Insoluble (soluble in ethanol, ethyl ether, CCl₄, miscible with organic solvents) pН **VAPOR DENSITY REFRACTIVE INDEX** 1.441 - 1.443 FLASH POINT 94 C

APPLICATION

1.Diels-Alder Reaction

The [4+2]-cycloaddition of a conjugated diene and a dienophile (an alkene or alkyne), an electrocyclic reaction that involves the 4 π -electrons of the diene and 2 π -electrons of the dienophile. The driving force of the reaction is the formation of new σ -bonds, which are energetically more stable than the π -bonds. In the case of an alkynyl dienophile, the initial adduct can still react as a dienophile if not too sterically hindered. In addition, either the diene or the dienophile can be substituted with cumulated double bonds, such as substituted allenes. With its broad scope and simplicity of operation, the Diels-Alder is the most powerful synthetic method for unsaturated six-membered rings. A variant is the hetero-Diels-Alder, in which either the diene or the dienophile contains a heteroatom, most often nitrogen or oxygen. This alternative constitutes a powerful synthesis of six-membered ring heterocycles. (source: http://www.organic-chemistry.org/)

2. Inhibitor of oxidative phosphorylation \rightarrow pesticide

Some insecticides inhibit or disrupt energy production. Initially, the insect can mobilize enough stored energy to continue its basic functions. While it can eat and digest food in the initial stages after being poisoned, it cannot produce more energy from the food. Eventually, the insect "runs out of steam," stops eating and even moving, and dies. (source: <u>http://www.entmclasses.umd.edu/</u>)

Acetylenedicarboxylate compounds		
Product	CAS RN.	
2-Butynedioic acid	142-45-0	
Cellocidin	543-21-5	
Diethyl 2-butynedioate	762-21-0	
Methyl acetylenedicarboxylate	762-42-5	
Potassium hydrogen butynedioate	928-04-1	

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Dipropargyl acetylenedicarboxylate	3154-91-4
Di-tert-butyl 2-butynedioate	66086-33-7
Bis(triphenyltin) acetylenedicarboxylate	73940-87-1

STABILITY AND REACTIVITY				
STABILITY	Stable under normal conditions.			
CONDITIONS OF INSTABILITY				
INCOMPATIBLE MATERIALS	Strong oxidizing agents, Strong acids, Strong bases, Strong reducing agents.			
DECOMPOSITION PRODUCTS	Carbon monoxide, Carbon dioxide.			
POLYMERIZATION	Will not occur			

SAFETY

HAZARD NOTES	Causes burns. Material is extremely destructive to tissue of the mucous membranes and upper respiratory tract, eyes, and skin. Inhalation may result in spasm, inflammation and edema of the larynxand bronchi, chemical pneumonitis, and pulmonary edema. Symptoms of exposure may include burning sensation, coughing, wheezing, laryngitis, shortness of breath, headache, nausea, and vomiting. Exposure can cause: Nausea, headache, and vomiting. Cyanosis. Convulsions.
EYE	Material is extremely destructive to eye.
SKIN	Material is extremely destructive to skin.
INGESTION	Material is extremely destructive to tissue of the mucous membranes and upper respiratory tract.
INHALATION	May result in spasm, inflammation and edema of the larynxand bronchi, chemical pneumonitis, and pulmonary edema
CHRONIC	
NFPA RATING	Health: 3, Flammability: 2, Reactivity: 0

SALES SPECIFICATION

APPEARANCE	colorless to light yellow liquid
ASSAY	98.0% min
REFRACTIVE INDEX	1.446 - 1.448

TRANSPORT & REGULATORY INFORMATION

UN NO.	3265
HAZARD CLASS	8
PACKING GROUP	
HAZARD SYMBOL	С
RISK PHRASES	34
SAFETY PHRASES	23-26-27-36/37/39-45

PACKING

PRICE